

**International General Certificate of Secondary Education
CAMBRIDGE INTERNATIONAL EXAMINATIONS**

PHYSICAL SCIENCE

0652/5

PAPER 5 Practical Test

CONFIDENTIAL INSTRUCTIONS

OCTOBER/NOVEMBER SESSION 2002

1 hour 30 minutes

Great care should be taken that this confidential information does not reach the candidates either directly or indirectly.

The Supervisor's attention is drawn to the form on page 4 which must be completed and returned with the scripts.

Instructions for preparing apparatus

These instructions detail the apparatus and reagents required by each candidate for each experiment in this paper. A summary of the questions that will be presented to the candidates is included, where appropriate, to allow the teacher to test the apparatus appropriately. **No access is permitted to the question paper in advance of the examination session.**

It is assumed that the ordinary apparatus of a science laboratory will be available, including a supply of purified water (distilled or deionised).

If arrangements are made for different sessions for different groups of candidates, care must be taken to ensure that the different groups of candidates are effectively isolated so that **no information passes between them.**

For Question 1

Each candidate will require:

- (i) a glass 100 cm³ beaker;
- (ii) a 250 cm³ beaker;
- (iii) access to a balance capable of reading to an accuracy of 0.1 g;
- (iv) a 0–110 °C thermometer in 1 °C steps;
- (v) ammonium chloride labelled 'solid Z'; approximate mass between 3.5 g and 4.0 g;
- (vi) a 50 cm³ measuring cylinder;
- (vii) about 30 cm³ water, coloured with addition of a simple dye, e.g. food colouring, labelled 'liquid L';
- (viii) a Bunsen burner.

For Question 2

Each candidate will require:

- (i) a 200 cm³ or 250 cm³ conical flask;
- (ii) a 50 cm³ or 100 cm³ measuring cylinder;
- (iii) a 10 cm³ measuring cylinder;
- (iv) a 0–110 °C thermometer;
- (v) a Bunsen burner;
- (vi) a stop clock;
- (vii) a sheet of clean paper A5 size;
- (viii) 300 cm³ of solution **H** which is a solution of sodium thiosulphate of concentration 0.05 mol dm⁻³.
If using Na₂S₂O₃·5H₂O, this would be 12.4 g dm⁻³ of Na₂S₂O₃·5H₂O in each dm³;
- (ix) 30 cm³ of solution **J** which is hydrochloric acid of concentration 2.0 mol dm⁻³.

All solutions should be freshly prepared in bulk and distributed just prior to the examination.

Information required from the Supervisor:

The Supervisor is asked to carry out the experiments and to enter the results on a spare copy of the examination paper, clearly marked 'Supervisor's Results' and showing the Centre number. This should be returned with the scripts. Failure to do so may cause the candidates to be penalised.

This form must be completed and returned in the envelope with the scripts together with the seating plan and the Supervisor's Results mentioned below.

October/November 2002

General

The Supervisor is invited to give details of any difficulties experienced by particular candidates giving their names and candidate numbers. These should include reference to:

- (a) difficulties due to faulty apparatus;
- (b) accidents to apparatus or materials;
- (c) physical handicaps, e.g. short sight, colour blindness;
- (d) any other information that is likely to assist the Examiner, especially if this cannot be discovered in the scripts;
- (e) any help given to a candidate.

The Supervisor is asked to supply the following information:

Plan of work benches, giving details by candidate numbers of the places occupied by the candidates for each session and a copy of the 'Supervisor's Results'.

NAME OF CENTRE

SIGNED

Supervisor

CENTRE NUMBER

DECLARATION (to be signed by the Principal)

The preparation of the practical examination has been carried out so as to maintain fully the security of the examination.

NAME

(in block capitals)

SIGNED(Principal)